

**UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**  
International General Certificate of Secondary Education

## **MARK SCHEME for the October/November 2008 question paper**

### **0620 CHEMISTRY**

**0620/05**

Paper 5 (Practical Test), maximum raw mark 40

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Page 2	Mark Scheme	Syllabus	Paper
	IGCSE – October/November 2008	0620	05

1 Experiment 5 observations

rapid/vigorous o.w.t.t.e (1)      bubbles/fizz (1)  
 blue precipitate (1) [max 2]

Table of results

initial boxes correctly completed (1)  
 final boxes correctly completed (1)  
 differences correctly completed (1)      signs correct (1)  
 comparable to Supervisor's results (1) [5]

(a) all 4 bars correctly drawn (3), –1 for each incorrect  
 labelled (1) [4]

(b) (i) solid **A**/Experiment 1 (1) [1]

(ii) temperature increased/heat/energy given out (1) [1]

(c) Experiment 3/solid **C** or from results (1) [1]

(d) (i) double the value or  $2 \times$  value from table (1) [1]  
 e.c.f. applies

(ii) half the value or half the value from table (1) [1]

(iii) more/larger volume of water (1) twice as much (for solid to dissolve in) o.w.t.t.e. (1) [2]

(e) acid present (1) carbonate present (1) carbon dioxide (1) [max 2]

**[Total: 20]**

2 (a) solution **K**      blue/green  
 solution **L**      yellow/orange/brown (1) [1]

(b) check pHs from Supervisor's results  
 pH of solution **K**      1–6  
 pH of solution **L**      1–6 (1) [1]

tests on solution **K**

(c) (i) blue (1) precipitate (1) [2]

(ii) deep/royal blue (1) solution/precipitate dissolves (1) [2]

(iii) no reaction/change/nothing (1) not lighter [1]

(iv) white (1) precipitate (1) [2]

Page 3	Mark Scheme	Syllabus	Paper
	IGCSE – October/November 2008	0620	05

tests on solution L

- (d) (i) (rusty) brown (1) precipitate (1) [2]  
(ii) brown precipitate (1) not red [1]  
(iii) no reaction/change/nothing (1) [1]  
(iv) white (1) precipitate (1) not lighter/paler [2]
- (e) acidic (1) [1]
- (f) chloride/  $Cl^-$  (1) [1]
- (g) (i) copper (1) [1]  
(ii) iron (1) (III) (1) [2]

[Total: 20]

[Total for paper: 40]